The embodiments of the invention in which an exclusive property or privilege is claimed are defined as follows:

- 1. A method for synthesizing an intermetallic negative electrode for an electrochemical cell comprising: dissolving one or more metal salts in an organic solvent forming a solution; adding a metal reducing agent to the solution to form an intermetallic compound; precipitating the intermetallic compound; separating the precipitate from the solution and forming the precipitate into an electrode.
 - 2. A method according to claim 1, in which two or more metal salts are used.
- 3. A method according to claim 1, in which the metal salts are metal chlorides.
- 4. A method according to claim 3, in which the metal chlorides are selected from the transition metal elements and one or more of the group IIIa, IVa and Va elements.
- 5. A method according to claim 3, in which the metal chlorides are two or more of Al, In, Sn, Sb, Cr, Mn, Fe, Co, Ni, Cu and Zn.
- 6. A method according to claim 1, in which the organic solvent is selected from one or more of ethylene glycol, ethanol, glycerol and xylenes.
 - 7. A method according to claim 1, in which the reducing metal is Zn or Mg.
- 8. A method according to claim 1, in which the intermetallic electrode contains an excess of one or more of metal element components within the electrode matrix.

- 9. A method according to claim 1, in which stoichiometric amounts of metal salts are dissolved in an organic solvent, stirred for ½ to 10 h at approximately 0 °C during the addition of a reducing metal powder, and then stirred at room temperature for another 2 to 24 h.
- 10. A method according to claim 9, in which the solid intermetallic product is annealed in an inert atmosphere or in a reducing atmosphere at 150 400 °C.
- 11. A method according to claim 1, and further comprising adding one or more grain growth inhibitors, porosity regulators, polymeric binders and electronically conducting additives.
- 12. An electrochemical cell containing an intermetallic negative electrode, a non-aqueous electrolyte and a positive electrode, the intermetallic electrode being made in accordance with the method of claim 1.
- 13. A battery consisting of a plurality of cells of claim 12, said cells being arranged in series and/or in parallel.